

Environmental Effects on Fortified Palm Olein in the Anthropocene

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Abstract— Until now, no literature exists to explain about the impact of time, and environmental parameters, when using typical commercial grade transparent linear low density polyethylene (LLDPE) packaging for protecting vegetable cooking oil during typical shelf-life conditions. We tested long-term impact of temperature and illuminance (light: 100 to 150 lux and darkness <1 lux) on refined bleached deodorized palm olein (RBDPOL), fortified with vitamin A. Results showed degradation of fortified RBDPOL with 70 ppm vitamin A was approximately 15% at 18-22°C in Light, 10% at 18-22°C in Darkness, 19% at 32-33°C in light and 17% 32-33°C in darkness. Similar trend was observed at 45 ppm vitamin A. Exposure to heat and light impacted vitamin A degradation, but is primarily attributed to temperature, irrespective of packaging. This investigation builds on our previous work (Silalahi et al., 2017), and confirms the impact of challenging environmental conditions on vegetable cooking oil shelf-life over long time periods.

Key words --- Vitamin A, Shelf-life, Illuminance, Stability, Fortified Palm, Packaging,

I. INTRODUCTION

Providing sustainable food in the Anthropocene is never more urgent. Many environmental systems and processes are pressed because food production is not operating efficiently. Achieving healthy sustainable food systems will require large reductions in food loss and waste, together with improved food production practices [1]. Within this context, from 2012, several nations adopted fortification policies to include vegetable oil as a fortified delivery mechanism [2, 3]. More recently, one global supplier of palm oil, has also opted to fortify palm based vegetable cooking oil with vitamin A [4, 5].

Edible oil fortification, is a potentially useful means of expanding the present reach of vitamin A [6] But, stability is shown to be problematic in some environments [7, 5]. Factors such as acidity, moisture, ultraviolet light, oxygen and heat are well established causes [8 -12]. Clearly, given new global challenges of delivering sustainable food systems in the Anthropocene, attention should also be considered, not only for the food system itself, but also for the total delivery mechanism, naturally comprising materials used to protect the food system [1]. Without doubt, there is still continued interest to establish the most ideal methods for determining the most suitable environmental conditions and packaging materials [13] for long-term vitamin stabilisation [14, 15]. A recent study [1], noted several effects from choice of packaging material, to protect vitamin A and noted beneficial effects of Polyethylene Terephthalate (PET) vs transparent nylon, possibly attributed to improved gas transmission rates and water vapour permeability. The same study showed that opaque Nylon enabled better vitamin A retention, compared to transparent PET, because vitamin A is known to be more sensitive to UV light [16]. than oxygen attack [17, 18].

Although, others [5] previously considered the effects of differing packaging materials on vitamin A degradation, the experimental design did not clearly show the effect of illuminance (lux). Based on these aforementioned factors, no literature exists to explain the

effect of illuminance on fortified vegetable oil using commercial packaging materials during typical environmental shelf-life conditions. Therefore, this study aims to establish this effect at several isothermal conditions using commercial food grade packaging.

As in previous studies, Refined Bleached Deodorised Palm Olein (RBDPOL) was used and preferred for its good oxidative stability [19, 5].

II. MATERIALS AND METHOD

A. RBDPOL

Supplied from PT. SMART, Tbk, Indonesia. Key parameters are described in Table 1. Vitamin A 1.0 mio IU, was supplied from DSM Nutritional Products Ltd., Singapore (Table 2).

TABLE 1.
REFINED BLEACHED DEODORIZED PALM OLEIN (RBDPOL)

Analysis Parameter	Method	Result
Iodine Value	AOCS Cd 1b-87	60.67±0.15
Peroxide Value	AOCS Cd 8-53	0.5±0.02
Free Fatty Acid	AOCS Ca 5a-40	0.04±0.00
Moist & impurities	AOCS Ca 2b-38	0.05±0.01
Color	AOCS Cc 13e-92	2.2±0.01

TABLE 2.
DESCRIPTION OF ESSENTIAL PARAMETERS OF VITAMIN A ASSAY

Parameter	Specification
Peroxide Value, meq kg ⁻¹	Max. 10.0
Acid value, mg KOH g ⁻¹	Max. 2.0

Assay	1.0–1.1 mio IU g ⁻¹
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B. Iodine Value (IV), Peroxide value (PV), Free fatty acid (FFA), Moisture & Impurities (M&I), Colour measurement determination

The iodine value was determined by iodometric titration (AOCS Ca 1b-87). Peroxide value was determined by AOCS Cd 8-53, 2009. Free fatty acids were determined by AOCS Ca 5a-40. Moisture was determined by AOCS Official Method Ca 2b-38. Oil colour was determined by Lovibond Tintometer colour scale (AOCS Cc13e-92).

C. Storage conditions and sampling

Oil samples were prepared in typical laboratory conditions, then stored with or without light exposure, at two temperature ranges; first 18-22°C [20], representing typical air-conditioned supermarket and second 32-33°C [21] representing non-temperature control storage (Table 3). Light measurement (illuminance), was quantified using a Digital Lux Meter LX1010B (supplied by Mextech, India). Lux, being equal to one lumen per square metre (1 lx = 1 lm/m²). The JIS C 1609-1 standard [22, 23] defines performance requirements for illuminometers. In this study darkness is defined as <1 Lux. Illuminance (light) = 100 to 150 Lux.

TABLE 3.

STORAGE CONDITIONS FOR FORTIFIED (VITAMIN A) RBDPOL

Fortification Dosage Vitamin A	Storage Condition	
	18-22°C*	32-33°C**
45 ppm	Light	Light
	Dark	Dark
70 ppm	Light	Light
	Dark	Dark

*Storage 18-22°C represents air-conditioned supermarket (Howell, 1997)

**Storage 32-33°C represents non-temperature control (BMKG Jakarta 1981-2010)

Dark Condition : illumination 0 Lux

Light Condition : illumination 100-150 Lux (warehouse, CIBSE

Recommendation 2012)

Preparations of RBDPOL and vitamin A 1.0 mio IU (retinyl palmitate) were assembled at concentrations of 70 IU g⁻¹ and 45 IU g⁻¹, described previously [24, 5] and studied in 225ml transparent nylon LLDPE (linear low density polyethylene) pouch supplied from Panverta, Indonesia. (Specification Nylon15 μ / adhesive / LLDPE 120 μ).

D. Quantification of vitamin A by high performance liquid chromatography (HPLC)

The vitamin A was determined by liquid chromatography method [25]. Standardized vitamin A (Retinyl palmitate) was supplied from Supelco-Sigma Aldrich (purity 93.8%).

HPLC determination was started with baseline check for a minimum of 30 minutes with mobile phase methanol: ultrapure water (95:5) at flow rate 0.5 mL/min. Instrument: Shimadzu Prominence-i LC-2030 C. Column: Shim Pack GIST C18 4.6 x 150 mm. Mobile phase: methanol: ultrapure water (95:5). Flow rate: 0.5 mL/min.

Vol. Inject: 10 μL. Detector: UV. Wavelength: 328 nm. Calculation: Determination of vitamin A in sample using calibration.

$$y = bx + a; x = \frac{(y - a)/b}{\text{weigh of sample (g)}} \quad (1)$$

Where: y = peak area

b = slope

a = intercept

x = Vitamin A IU/g in sample

Calculation of retained vitamin A in RBDPOL in n month was determined by the following:

$$\frac{C_n}{C_0} \times 100\% \quad (2)$$

C_0 = vitamin A content: 0 month (IU/g)

C_n = vitamin A content: n month (IU/g)

E. Statistical analysis

Statistical analysis was first examined using a combination of Student t-test to determine uncertainty [26]. Final analysis was carried out using analysis of variance ($\alpha = 0.05$) with MiniTab v.16.1.1 software.

III. RESULTS AND DISCUSSION

A. Calibration of Vitamin A

A calibration of vitamin A level concentration (IU/g) is shown previously in [1]. Calculations show corellation between peak area obtained from analysis using HPLC and regression of linearity.

B. Effect of temperature and illuminance on vitamin A stability

Figures 1 and 2, show the impact of both temperature and illuminance (light: 100 to 150 lux and darkness <1 lux) during storage of fortified RBDPOL with Vitamin A (70 and 45 ppm). After six (6) months monitoring, the degradation of RBDPOL fortified with 70 ppm vitamin A was approximately 15% at 18-22°C in Light, 10% at 18-22°C in Dark, 19% at 32-33°C in light and 17% 32-33°C in dark. A similar trend was observed in RBDPOL fortified with 45 ppm vitamin A, after 6 months. The vitamin A degraded by approximately 13% at 18-22°C in light, 13% at 18-22°C in dark, 22% at 32-33°C in light and 16% 32-33°C in darkness.

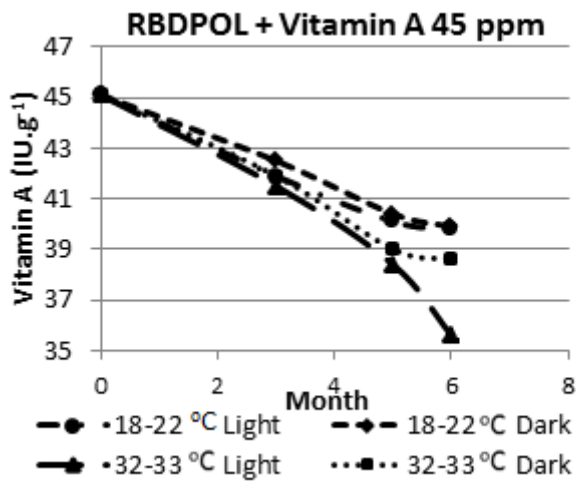


Fig. 1 Impact of temperature and illuminance (light: 100 to 150 lux and darkness <1 lux) during storage of fortified RBDPOL (45 ppm vitamin A)

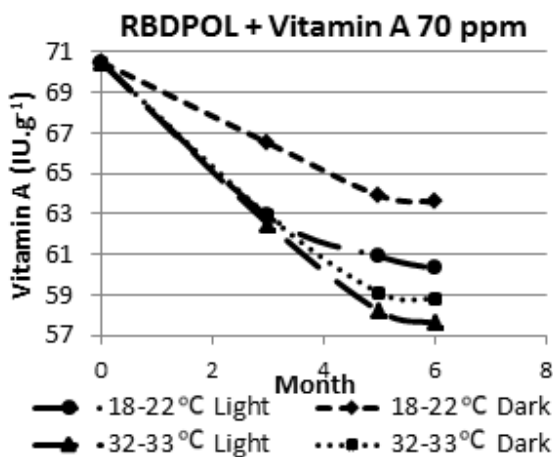


Fig. 2 Impact of temperature and illuminance (light: 100 to 150 lux and darkness <1 lux) during storage of fortified RBDPOL (70 ppm vitamin A)

The data also shows the effect of temperature on the degradation of vitamin A during storage. A temperature of 32-33°C giving higher Vitamin A degradation versus 18-22°C. This result confirms similar findings on temperature impact of vitamin A degradation in [12]. Secondly, although test samples did not have direct exposure to natural sunlight, this observation shows the effect of vitamin A fortified RBDPOL to direct light exposure (100 to 150 lux) or darkness (<1 lux). Light (100 to 150 lux) conditions, will accelerate vitamin A degradation in RBDPOL faster than if <1 Lux condition. Exposure of RBDPOL vegetable oil, does have impact on stability of vitamin A [10].

Table 4, shows that temperature storage is probably contributing the biggest impact to degradation of vitamin A. After six (6) months monitoring, all samples stored at 18-22°C had vitamin A degradation of approximately 10-15%. Whereas, all samples stored at 32-33°C degraded by approximately 16-22%. These data confirm similar findings [5].

C. Correlation of PV value to vitamin A degradation

As described previously [5], for vitamin A fortification studies, the peroxide value (PV) was monitored

to acceptable level of 10 meq kg⁻¹ [19]. Figures 3 and 4 show the effect of peroxide (PV) to decrease vitamin A. Linearity was observed for 45 ppm ($y = -0.3106x + 43.035$) and 70 ppm ($y = -0.603x + 67.149$). This confirms a similar tendency noted by others for a compound decrease of vitamin A in oxidized oil [5, 27, 28].

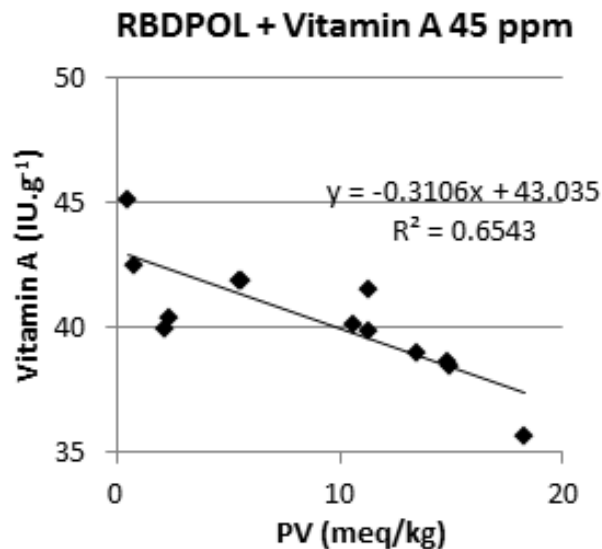


Fig. 3 Effect of PV during storage of fortified RBDPOL (45 ppm vitamin A)

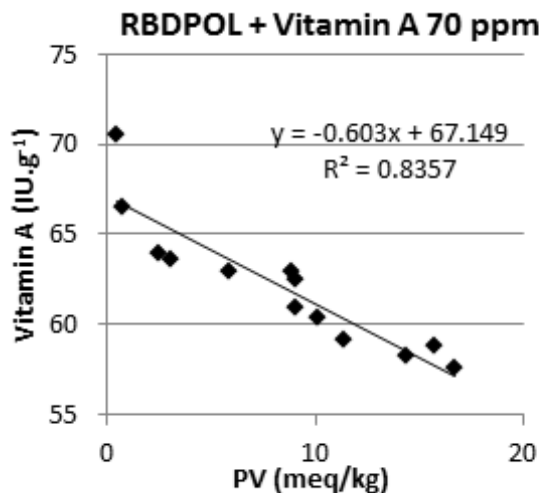


Fig. 4 Effect of PV during storage of fortified RBDPOL (70 ppm vitamin A)

Table 4.

Effect of Lux and temperature on degradation of fortified (vitamin A) RBDPOL

Vitamin A (ppm) into RBDPOL	Storage condition	MONTH										
		0			3			5			6	
		Vit A IU g ⁻¹	PV (meq/kg)	Vit A IU g ⁻¹	PV (meq/kg)	Vit A % Degraded	Vit A IU g ⁻¹	PV (meq/kg)	Vit A % Degraded	Vit A IU g ⁻¹	PV (meq/kg)	Vit A % Degraded
70ppm (Light**)	18-22 C	70.5±0.36	0.5±0.01	62.9±0.45	5.9±0.06	11	60.9±0.11	9.1±0.06	14	60.3± 0.03	10.1±0.05	15
70ppm (Dark#)		70.5±0.36	0.5±0.01	66.3±0.45	0.9±0.02	6	63.8±0.11	2.5±0.05	9	63.6± 0.20	3.1±0.05	10
45ppm (Light)		45.1±1.01	0.5±0.02	42.0±1.26	5.6±0.02	7	40.1±0.08	10.6±0.06	11	39.8± 0.14	11.3±0.06	13
45ppm (Dark)		45.1±1.01	0.5±0.02	42.5±0.01	0.8±0.06	6	40.4±0.37	2.4±0.01	10	39.9± 0.03	2.2±0.05	13
70ppm (Light)	32-33 C	70.5±0.36	0.5±0.01	62.5±1.16	9.2±0.09	11	58.3±0.06	14.4±0.04	17	57.6± 0.16	16.7±0.08	19
70ppm (Dark)		70.5±0.36	0.5±0.01	62.9±0.26	9.2±0.12	11	59.1±0.01	11.5±0.06	16	58.8± 0.11	15.7±0.06	17
45ppm (Light)		45.1±1.01	0.5±0.02	41.6±0.26	8.51±0.13	8	38.4±0.08	15±0.02	15	35.6± 0.04	18.3±0.01	22
45ppm (Dark)		45.1±1.01	0.5±0.02	41.9±0.75	5.5±0.13	7	39.0±0.02	13.5±0.04	14	38.6± 0.12	14.9±0.01	16

Dark = <1 Lux

** Light = 100 to 150 L

IV. CONCLUSION

Overall, after 6 months of sample monitoring, the degradation of vitamin A was more than 10% at the lowest test temperature. This study confirms previous experimental findings [5], that fortification 45 IU g⁻¹ in cooking oil is probably not enough to maintain the Vitamin A level at more than 20 IU g⁻¹ during the entire shelf life.

The impact of temperature has a dramatic effect during storage. Samples kept at the cooler condition (18-22°C), retained more vitamin A compared to samples kept at 31-33°C, clearly showing heat treatment strongly influenced the degradation vitamin A [5]. Additionally, this work shows the advantage of keeping fortified oil away from high illuminance, thereby minimizing degradation of vitamin A. Without doubt, the handling and storing of fortified vegetable oil, before consumption could positively or negatively influence the final content of vitamin A [14]. Cooler storage, low illuminance or low UV light adsorbent packaging could minimize excessive losses of vitamin A during distribution and storage [12]. While effective mandatory fortification will benefit from suitable manufacturing and distribution [14, 29], our results suggest there are benefits in selecting suitable packaging materials, which are able to minimize illuminance from natural or artificial UV light [15, 5], together with the most optimal storage temperature in order to maximize vitamin A retention. Introduction of synthetic or natural antioxidants, together with the aforementioned mitigation methods, is another variant not yet explored and requires investigation.

We believe, this information is important to product developers and nutritionists, especially when fortified vegetable oils are frequently allocated by the United Nations World Food Programme to challenging locations with inadequate environmental controls. With this thought, it is worth noting that many tropical countries, locally consume minimally processed red palm oil as an important source of provitamin A because of its high β -carotene content [1].

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